



by Maxim Mai

THERMODYNAMICS

“... the only physical theory of universal content concerning which I am convinced that, within the framework of the applicability of its basic concepts, it will never be overthrown.”



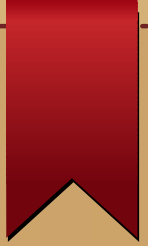
A. Einstein

THERMODYNAMICS

**** the laws ****

- I. When energy passes (work, heat, matter) into or out from a system, the system's internal energy changes in accord with the law of **conservation of energy**.
- II. In a natural thermodynamic process, the **sum of the entropies** of the interacting thermodynamic systems **increases**.
(+ two laws for consistency)

QUIZ

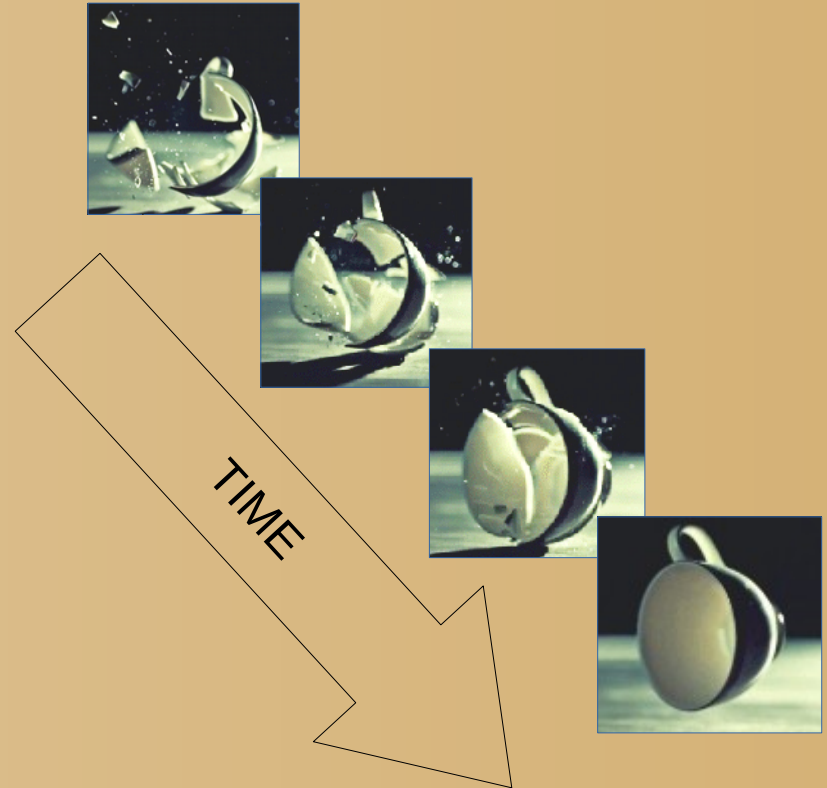


WHICH IS MORE NATURAL?

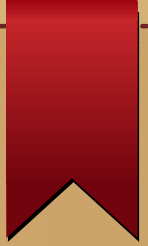
A



B

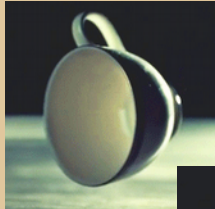


QUIZ

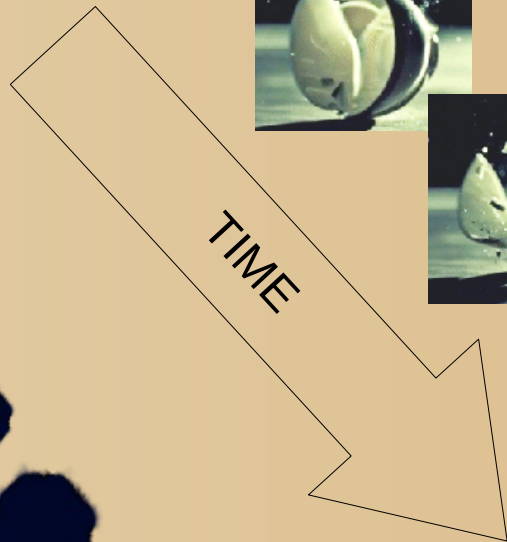


WHICH IS MORE NATURAL?

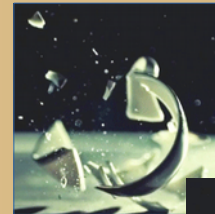
A



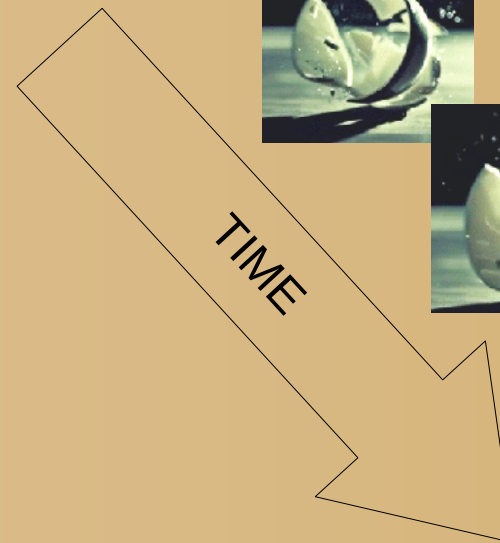
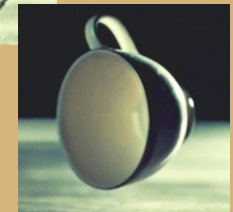
Entropy (S) in-creases
→ irreversible natural process



B

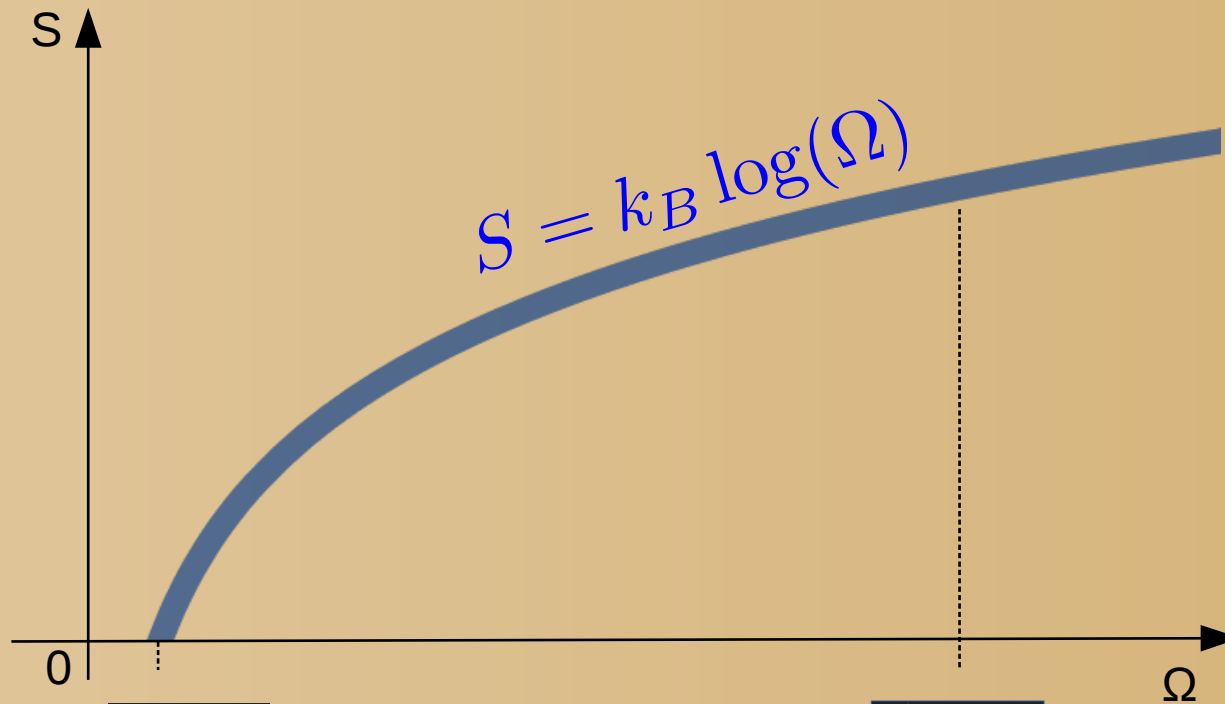


Entropy (S) de-creases
→ unnatural



ENTROPY

εν- 'in' and *τροπέ* 'turning, transformation'

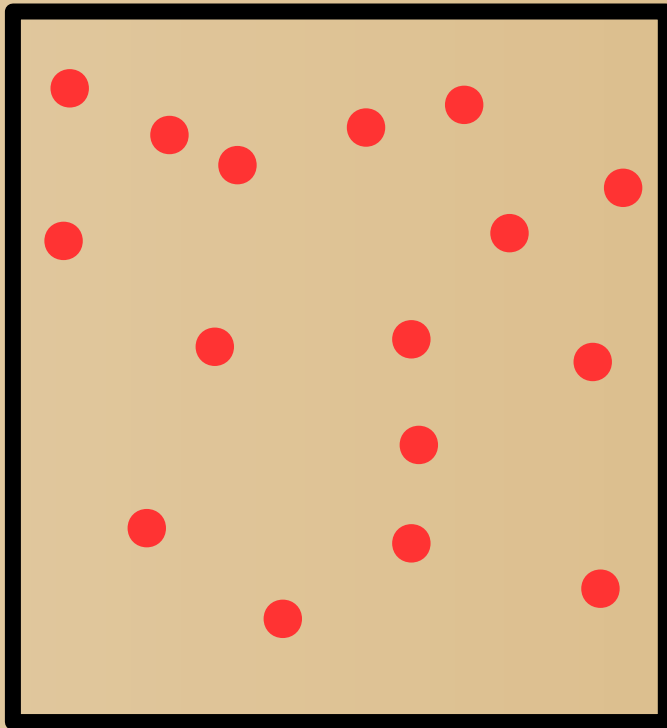


microscopic configurations of a system (microstates)

Many microstates = one macrostate

THERMODYNAMIC SYSTEMS

“enclosed system described by a set of variables”



VOLUME

spacial extension

TEMPERATURE

average kinetic energy

PRESSURE

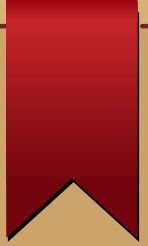
force per area

ENTROPY

...

CLASSICAL DEFINITION

(change of entropy)



$$\Delta S = \frac{\Delta Q}{T}$$

Heat (energy) transfer

Temperature



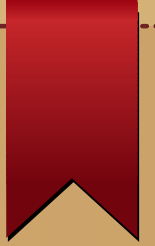
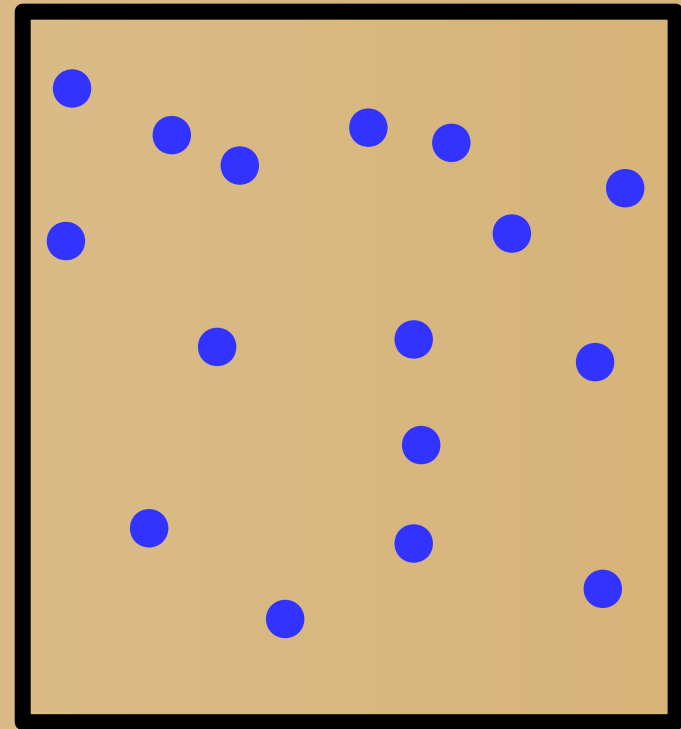
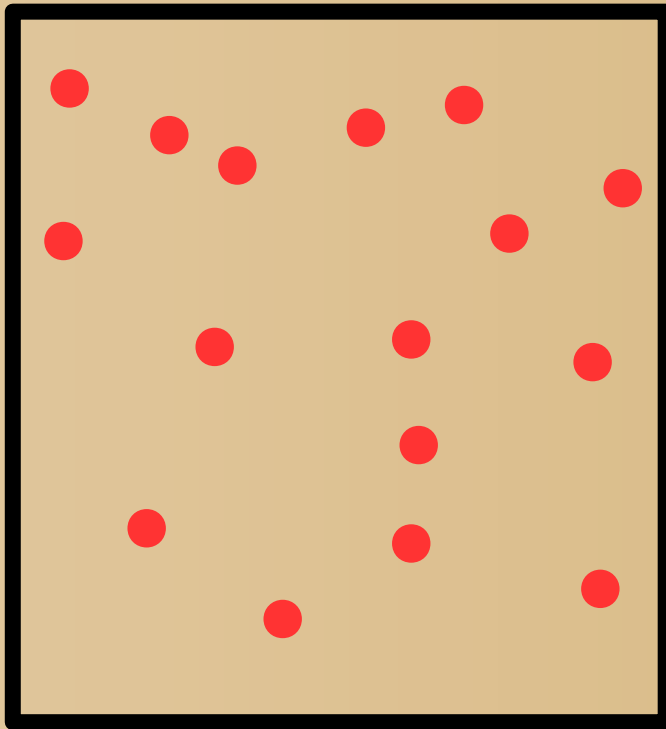
Clausius (1865)



CHANGE OF ENTROPY

Two isolated systems

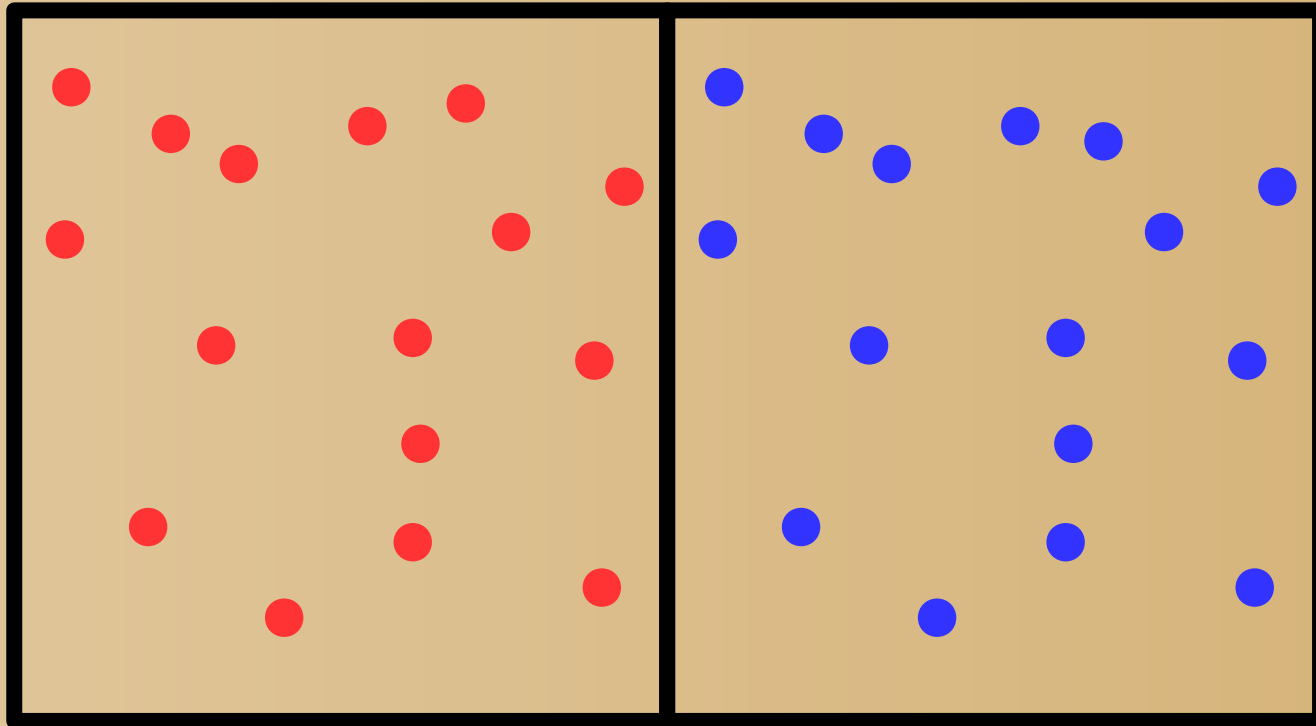
$$T_1 > T_2$$



CHANGE OF ENTROPY

Two systems in contact

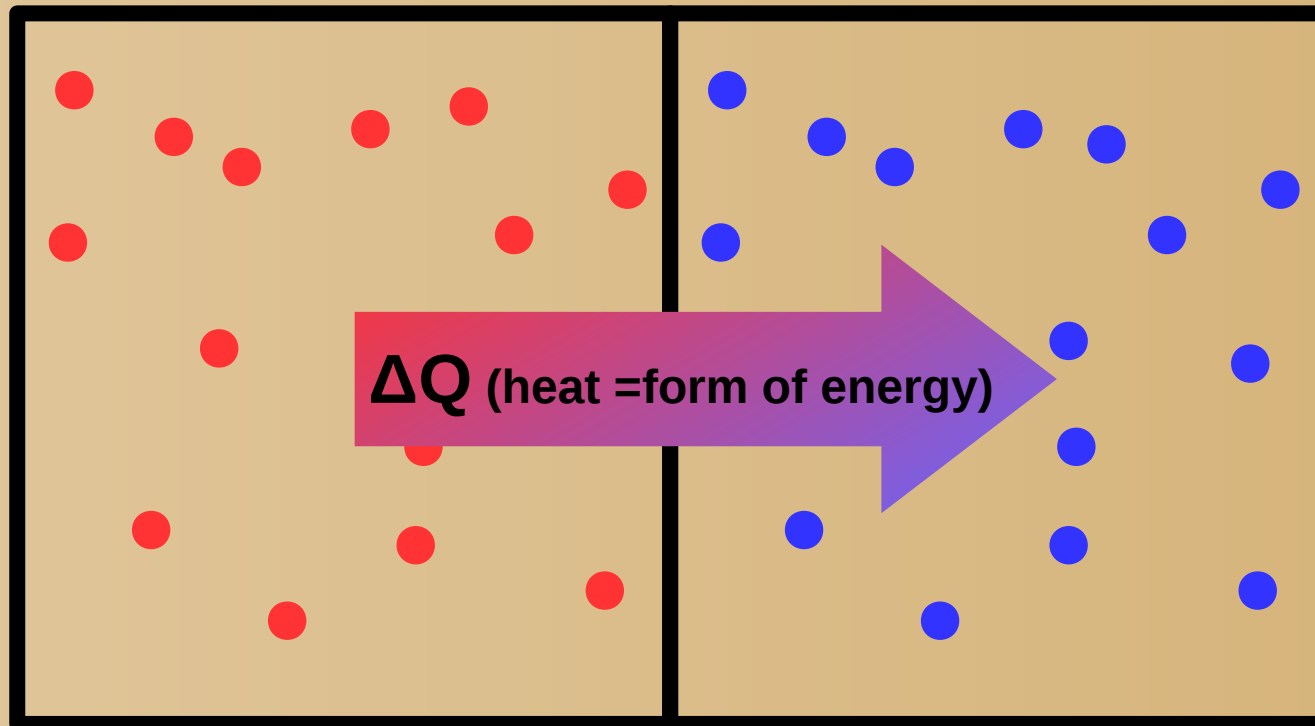
$$T_1 > T_2$$



CHANGE OF ENTROPY

Heat flow

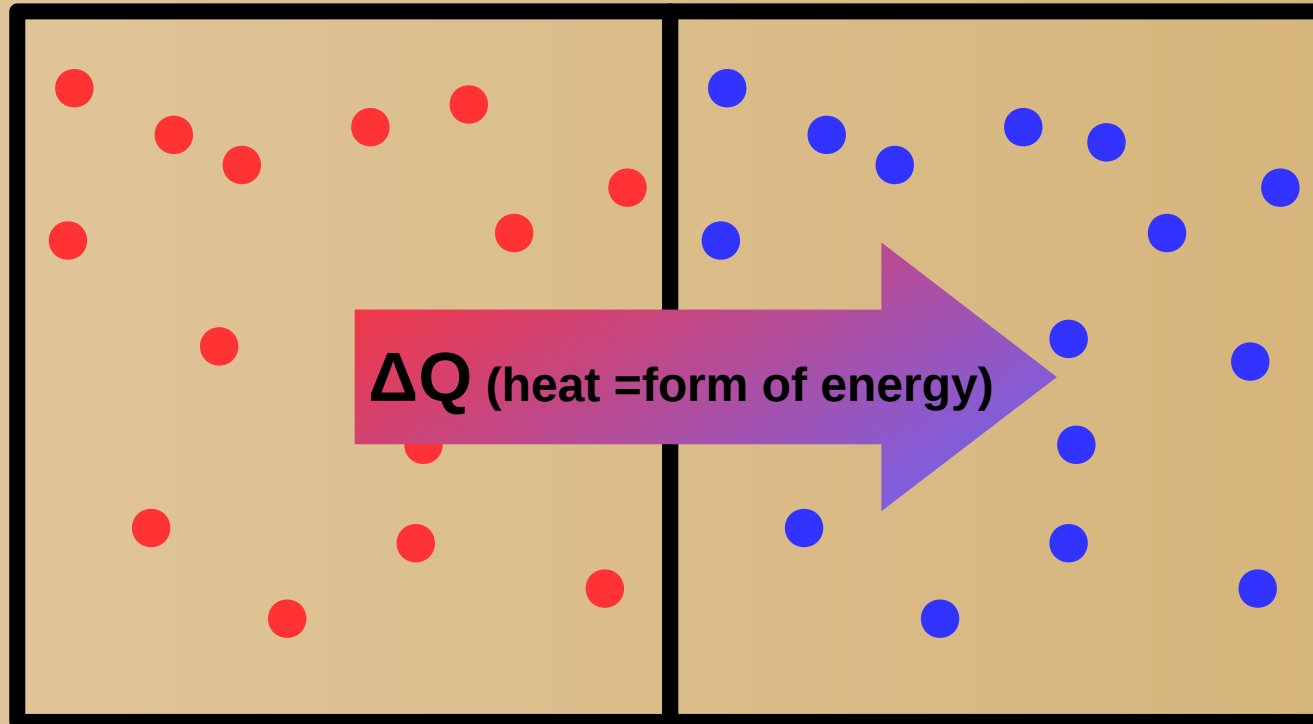
$$T_1 > T_2$$



CHANGE OF ENTROPY

Heat flow

$$T_1 > T_2$$

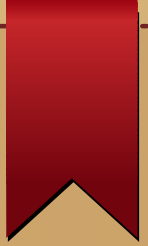


$$\Delta S_1 = -\frac{\Delta Q}{T_1}$$

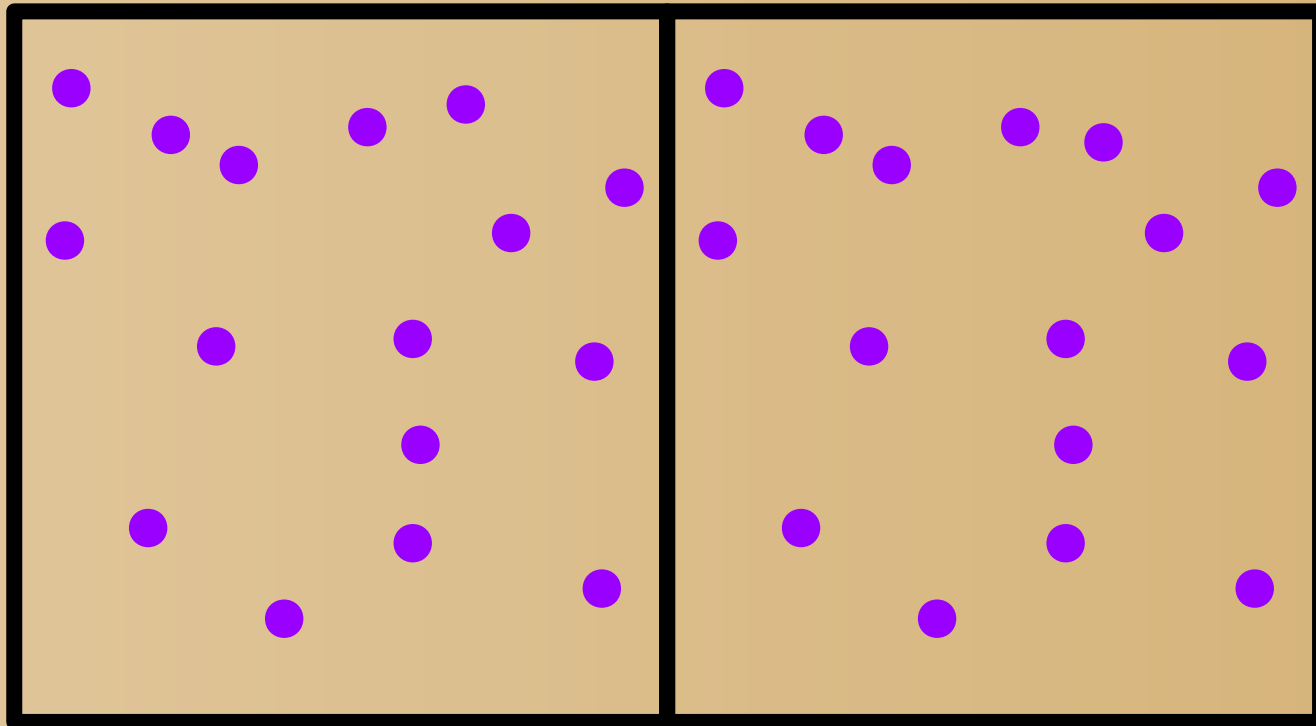
$$\Delta S_2 = +\frac{\Delta Q}{T_2}$$

CHANGE OF ENTROPY

equilibrium



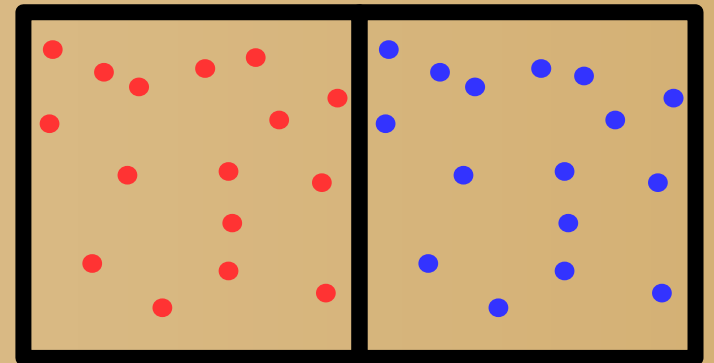
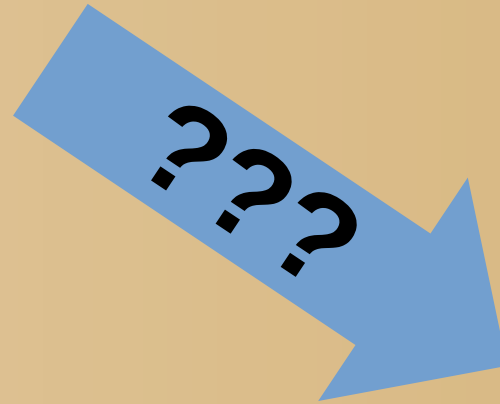
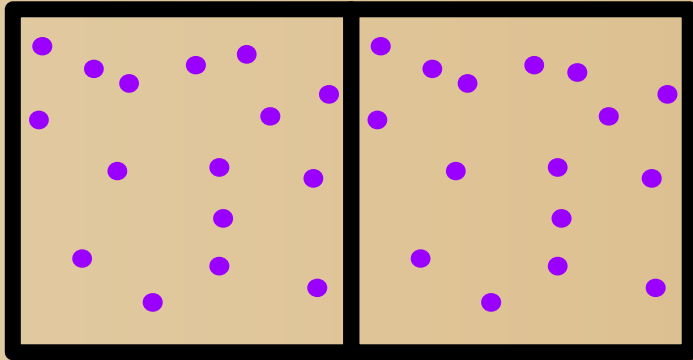
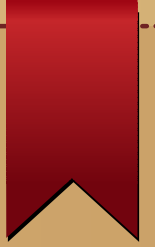
T_{Eq}



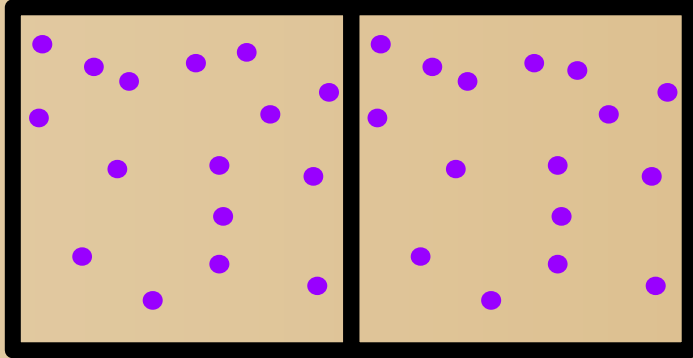
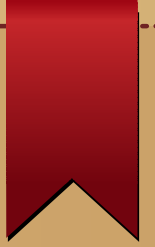
$$S_{Eq} > S_1 + S_2$$



QUIZ

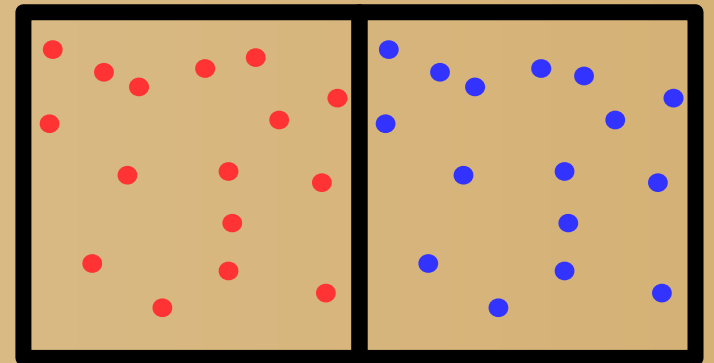
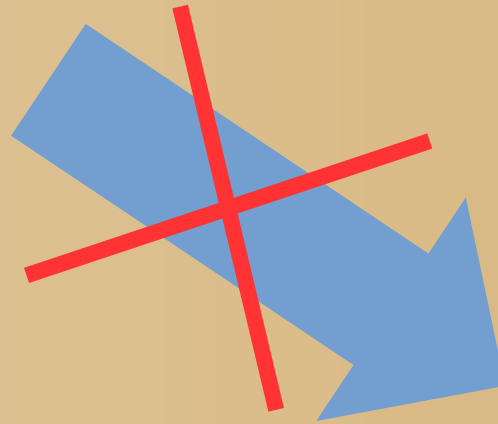


QUIZ

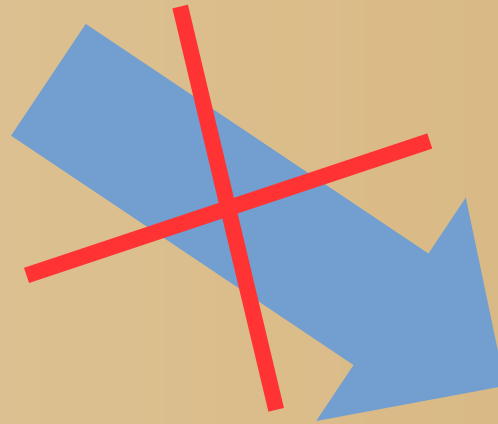
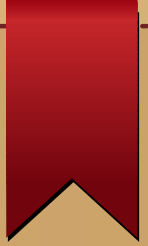


Overall entropy would decrease

→ **violates II. law**



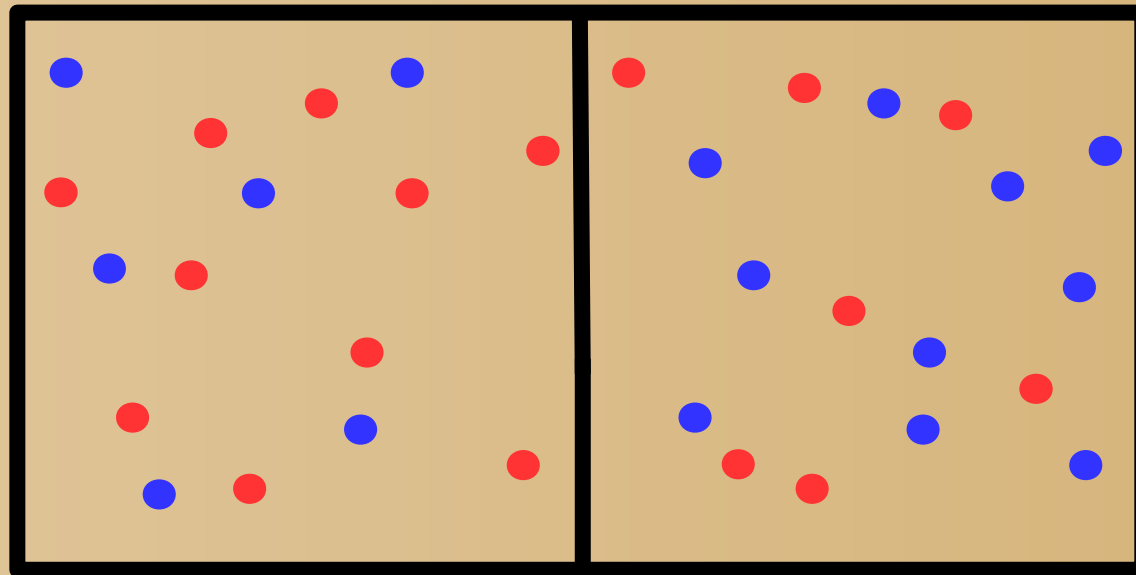
QUIZ



GEDANKEN-EXPERIMENT

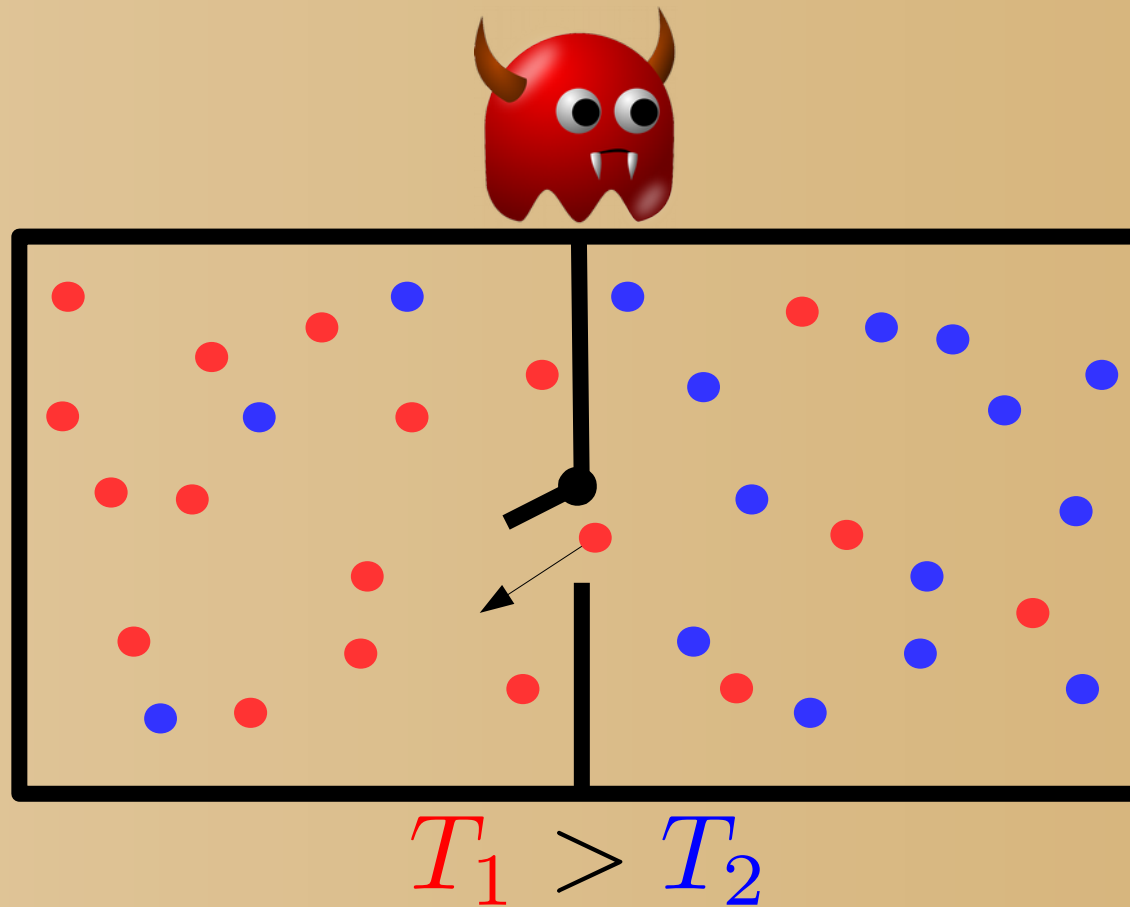
“Maxwell-demons” (1867)

$$T_1 = T_2$$



GEDANKEN-EXPERIMENT

“Maxwell-demons” (1867)



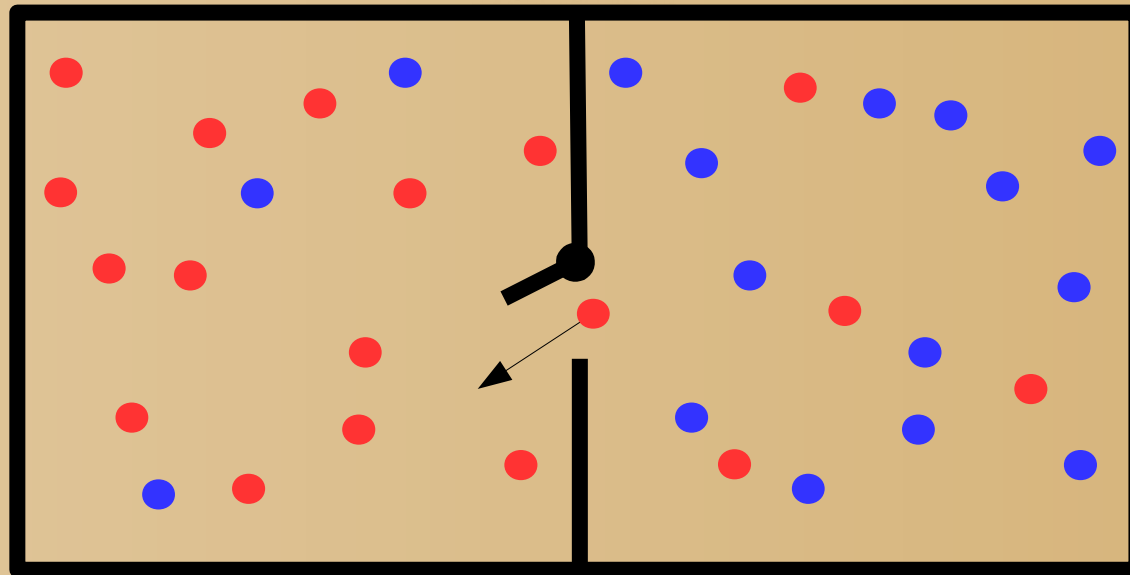
GEDANKEN-EXPERIMENT

“Maxwell-demons” (1867)

Maxwell-demon is part of the system.

- his entropy increases (demon heats up)
- the total entropy rises

II. LAW IS NOT VIOLATED



$$T_1 > T_2$$

thank you

